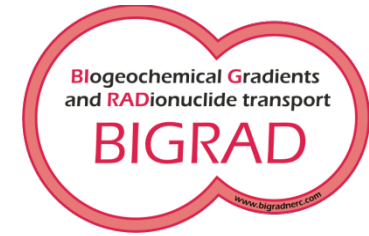
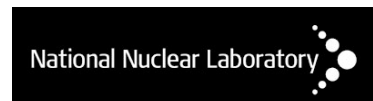
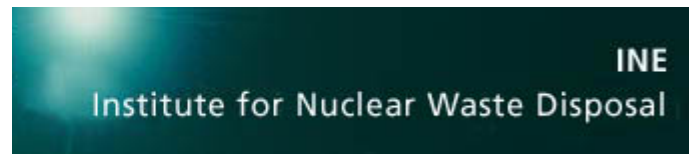


The NERC **BI**ogeochemical **G**radients and **RAD**ionuclide Transport **BIGRAD** Consortium



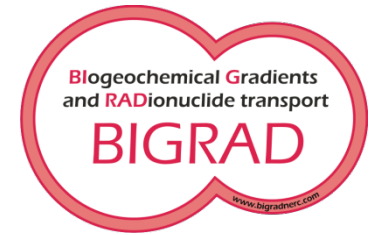
Katherine Morris

katherine.morris@manchester.ac.uk

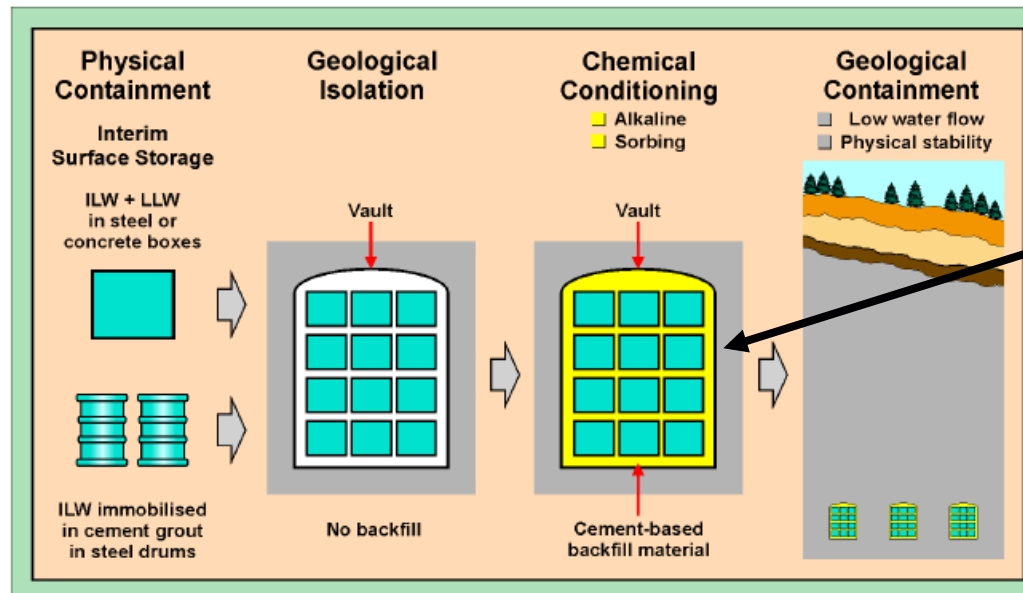




Radioactive Waste Disposal

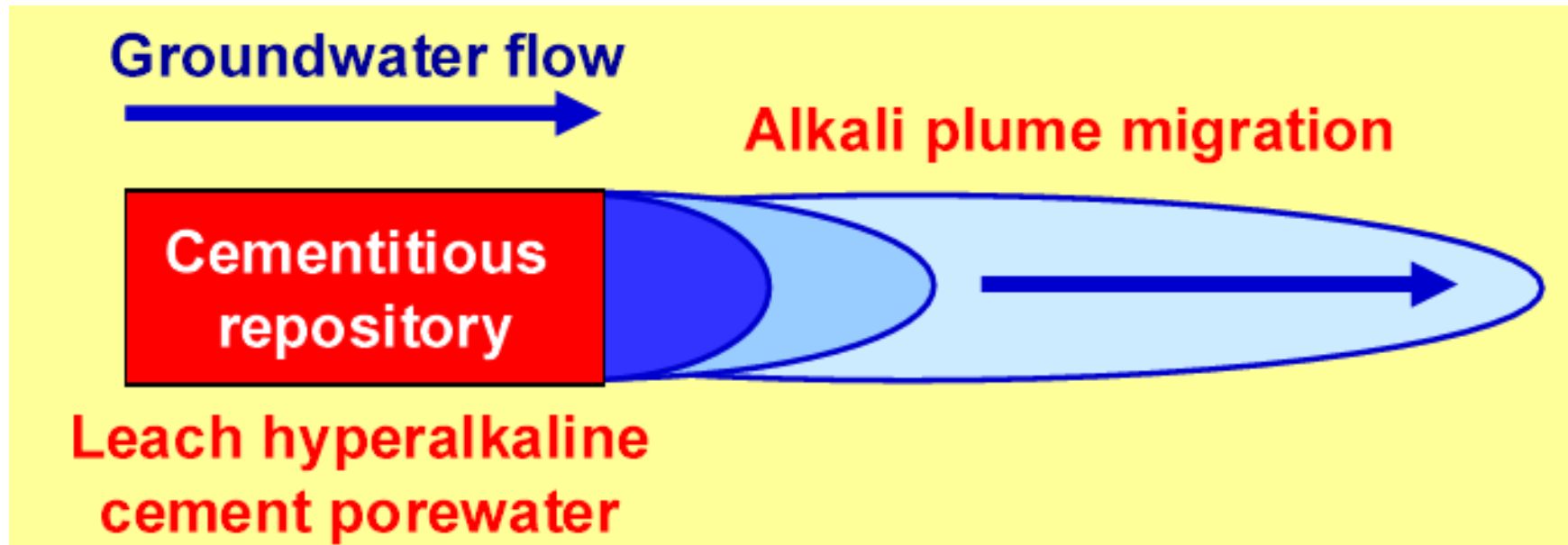
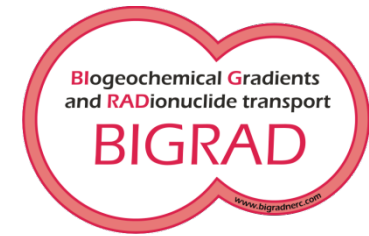


TheObserver



Chemically Disturbed Zone

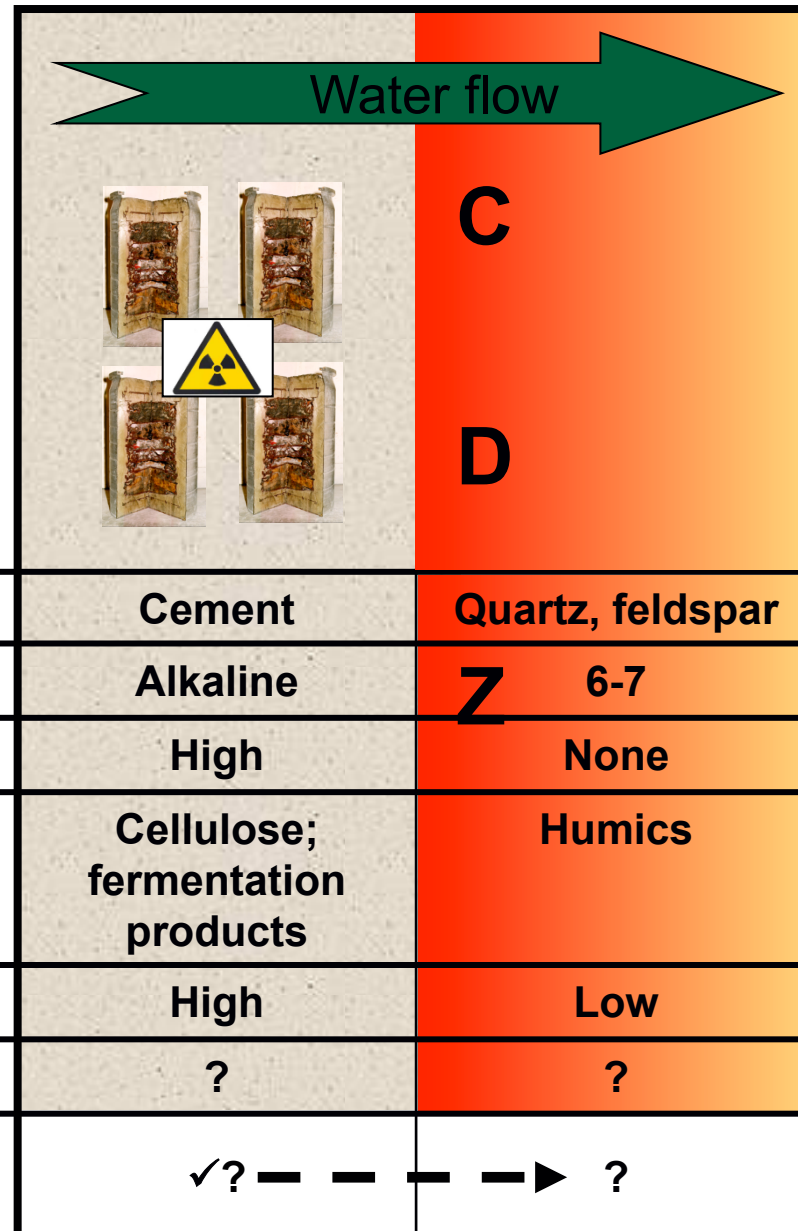
The Chemically Disturbed Zone



- CDZ is the dynamic interface between GDF and the natural environment
- CDZ impact on mineralogy / radionuclide transport largely unknown
 - Improved knowledge → improved safety case

The CDZ is interface between waste and natural environment:

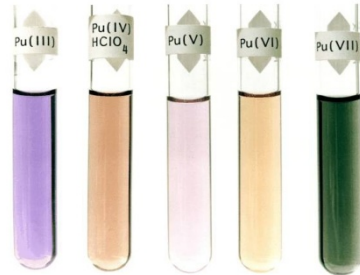
- steep, evolving biogeochemical gradients
- heterogeneous, nanometre to metre understanding required
- Tc, U, Np and Pu - evolution over 10^6 y



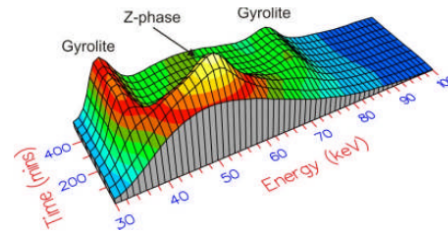
What is novel?



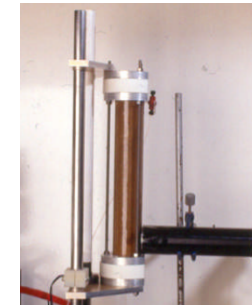
Biogeochemistry



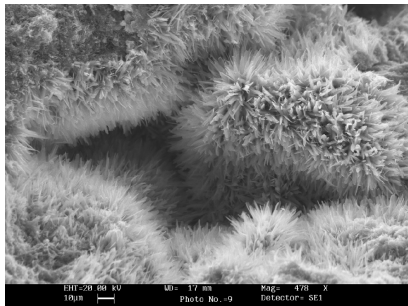
Solution chemistry



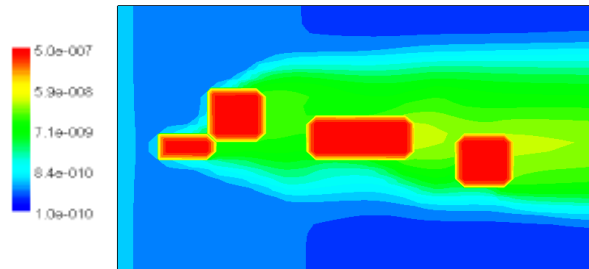
Mineral alteration



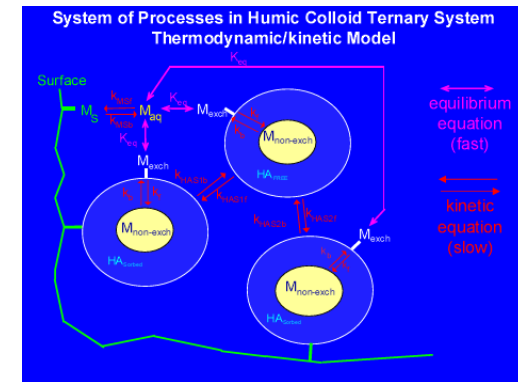
Transport



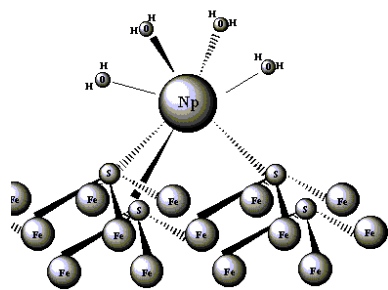
Porosity and permeability



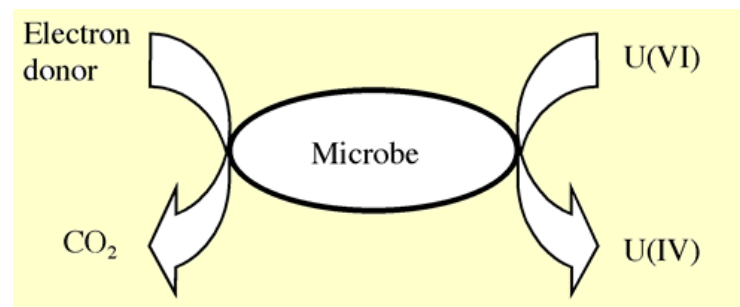
Multiscale predictive capability



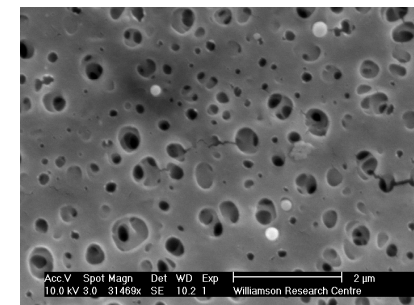
Sorption and reversibility



Radionuclide mineralogy

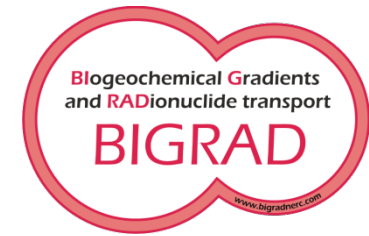


Radionuclide biogeochemistry



Colloidal behaviour

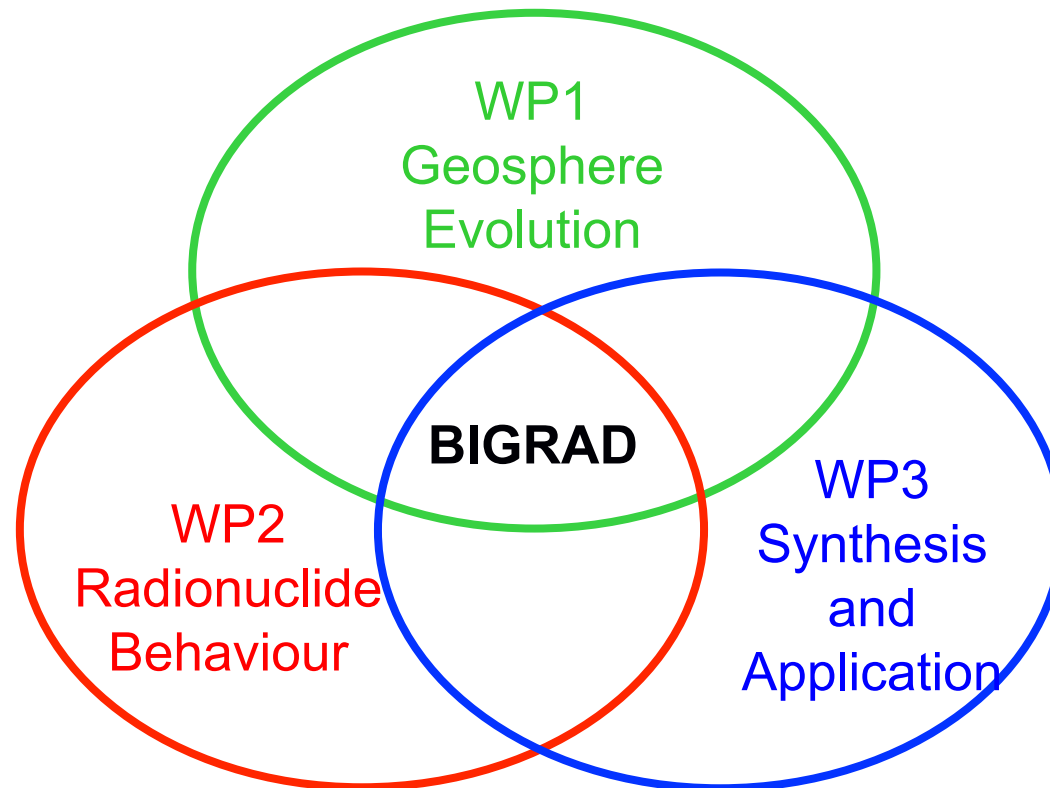
Scenarios / Systems



- GDF in acid / intermediate rock - likely candidate
- Minerals → mineral aggregates → model rock
- Acid / intermediate - quartz, feldspars, muscovite / chlorite / illite and iron oxides
- 4 year project, permeability, Permo-Triassic Sandstone
- Generic information relevant to GDF in a range of acid / intermediate geological settings

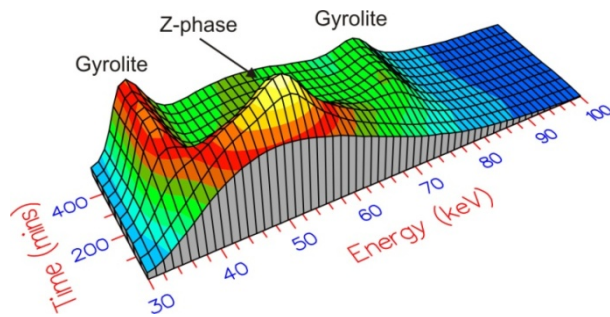
Research Hypothesis

The **biogeochemical gradients** that will develop across the poorly understood interface between an alkaline, deep geological disposal facility and the geosphere (the Chemically Disturbed Zone; CDZ) will be **critical controls on radionuclide behaviour and transport**, and thus on the **safety and environmental impact** of a geological disposal facility.



WP1 Geosphere Evolution

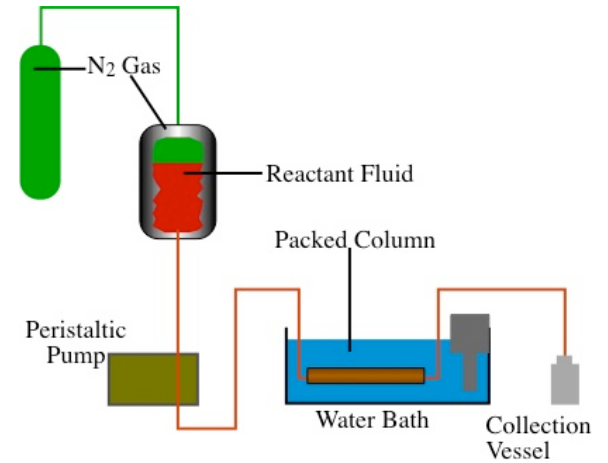
Mineral alteration



Geomicrobiology



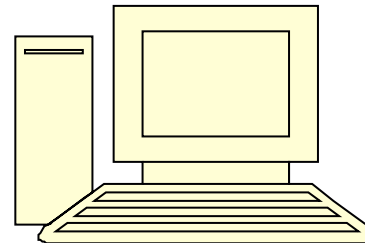
Physical processes



WP2 Radionuclide form reaction and transport



WP3 Synthesis and application to performance assessment



Nano-scale

Column-scale

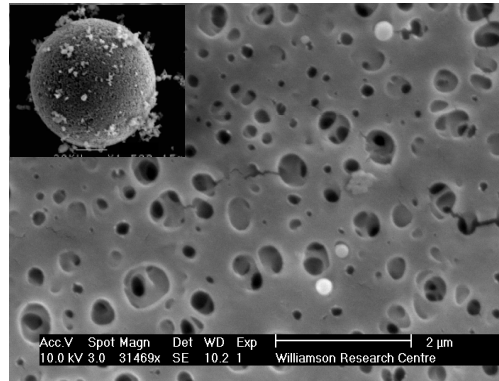


WP2 Radionuclide form, reaction and transport

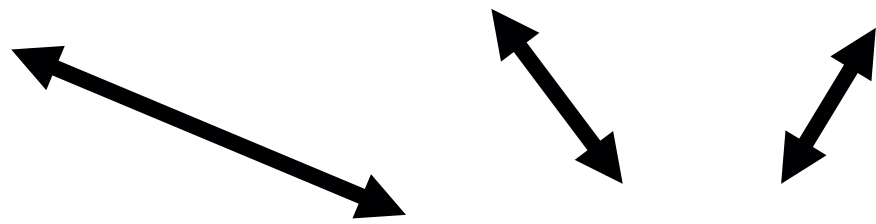
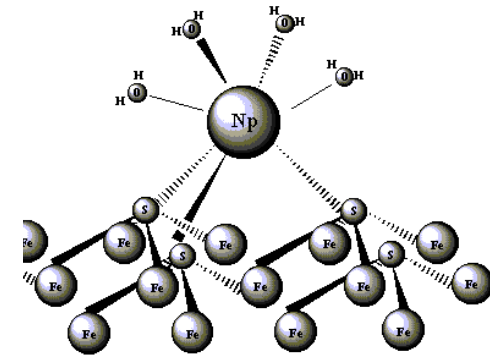
Aqueous speciation



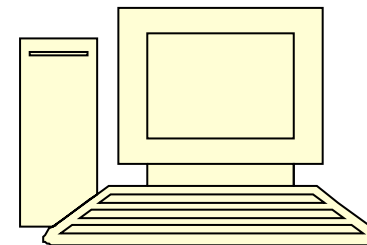
Colloids



Sorption and partitioning



WP3 Synthesis and application to performance assessment



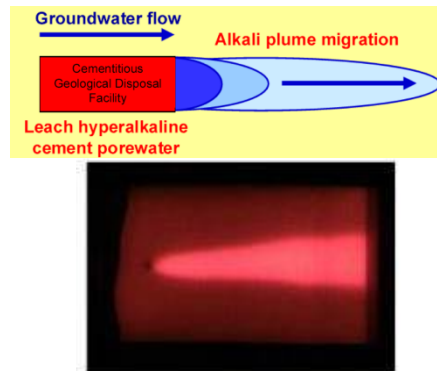
Nano-scale

Column-scale



WP3 Synthesis and application to performance assessment

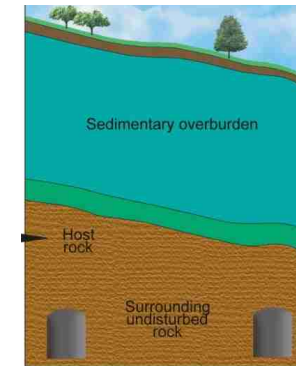
Conceptualisation,
experimental design &
bench-scale CDZ



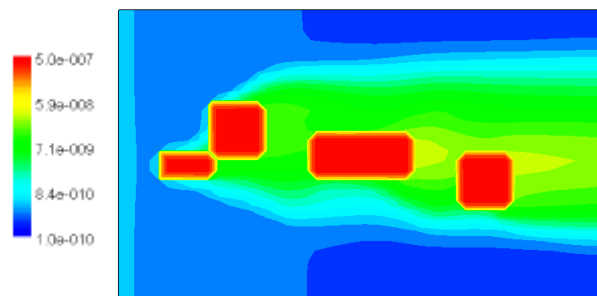
Process understanding,
parameterisation with
lab study data



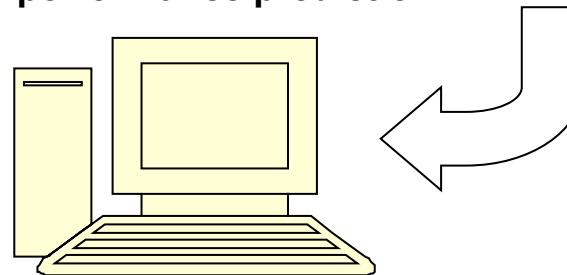
Variable GDF and
geosphere settings,
environmental conditions



Scenario modelling,
geosphere barrier function,
CDZ development

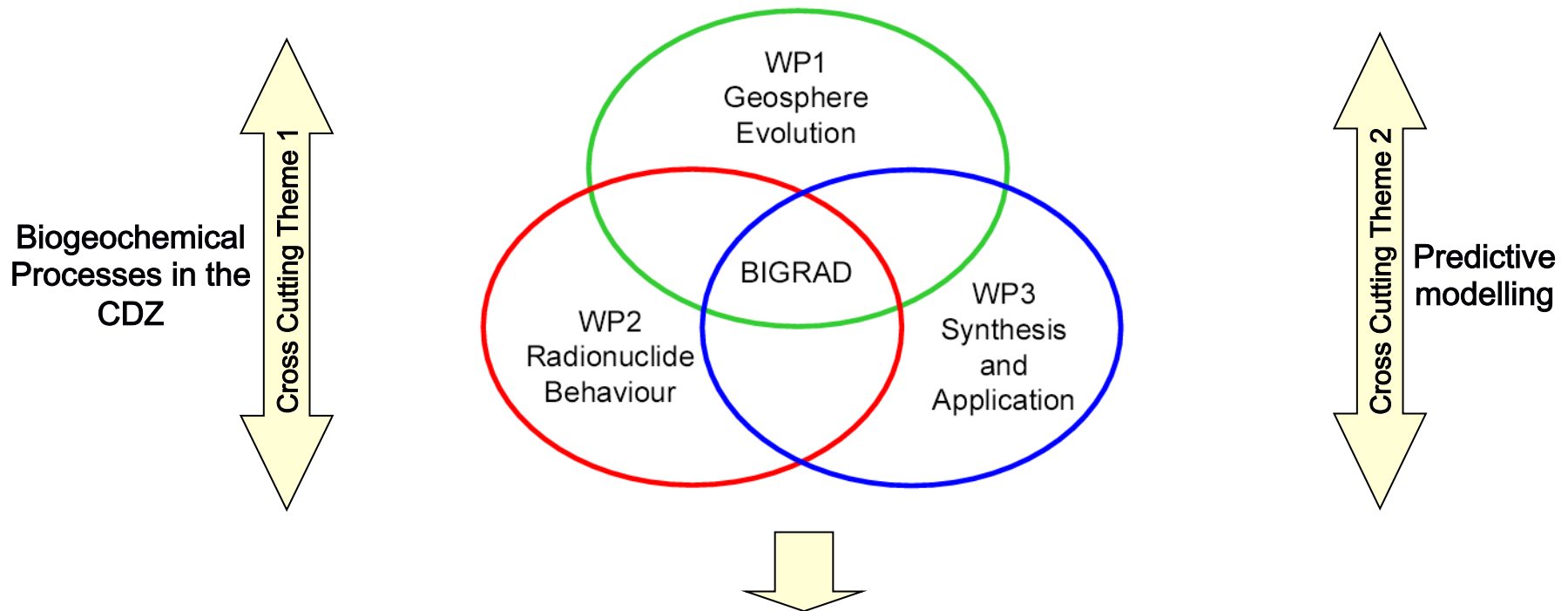


Numerical modelling tool
for performance prediction



Cross Cutting Themes

- CCT 1 Biogeochemical processes / radionuclide behaviour
- CCT 2 Predictive modelling capability / radionuclide mobility



Naturally integrated WPs and CCTs require a consortium

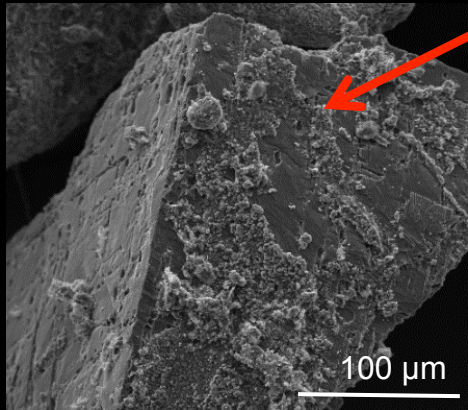


SCIENCE

katherine.morris@manchester.ac.uk

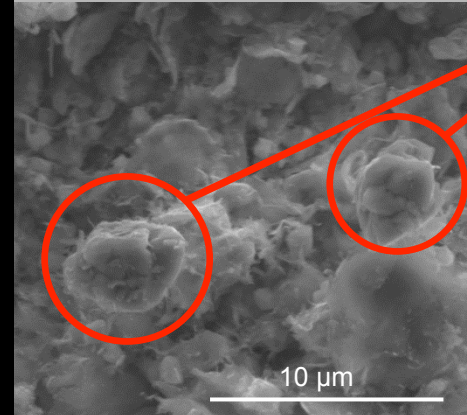
15 years BVG evolved fluid

Dolomite dissolution



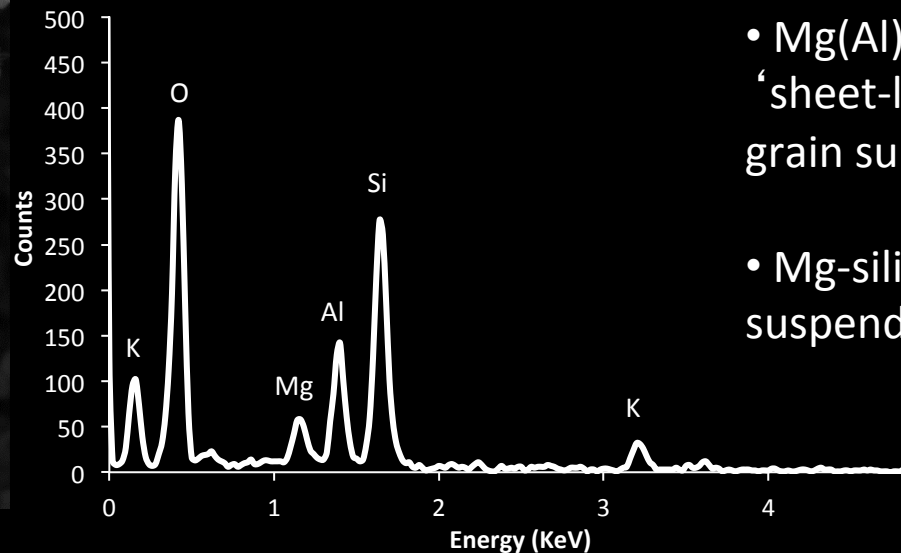
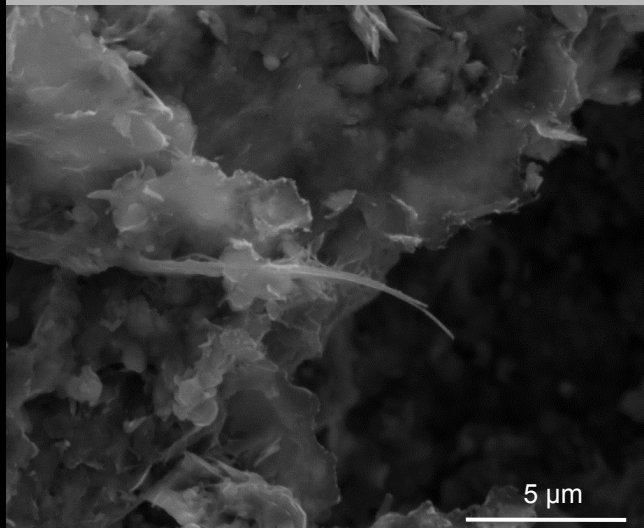
Dissolution features present on dolomite grains though less pronounced than in young fluid

Calcite precipitation



Calcite found on grain surfaces

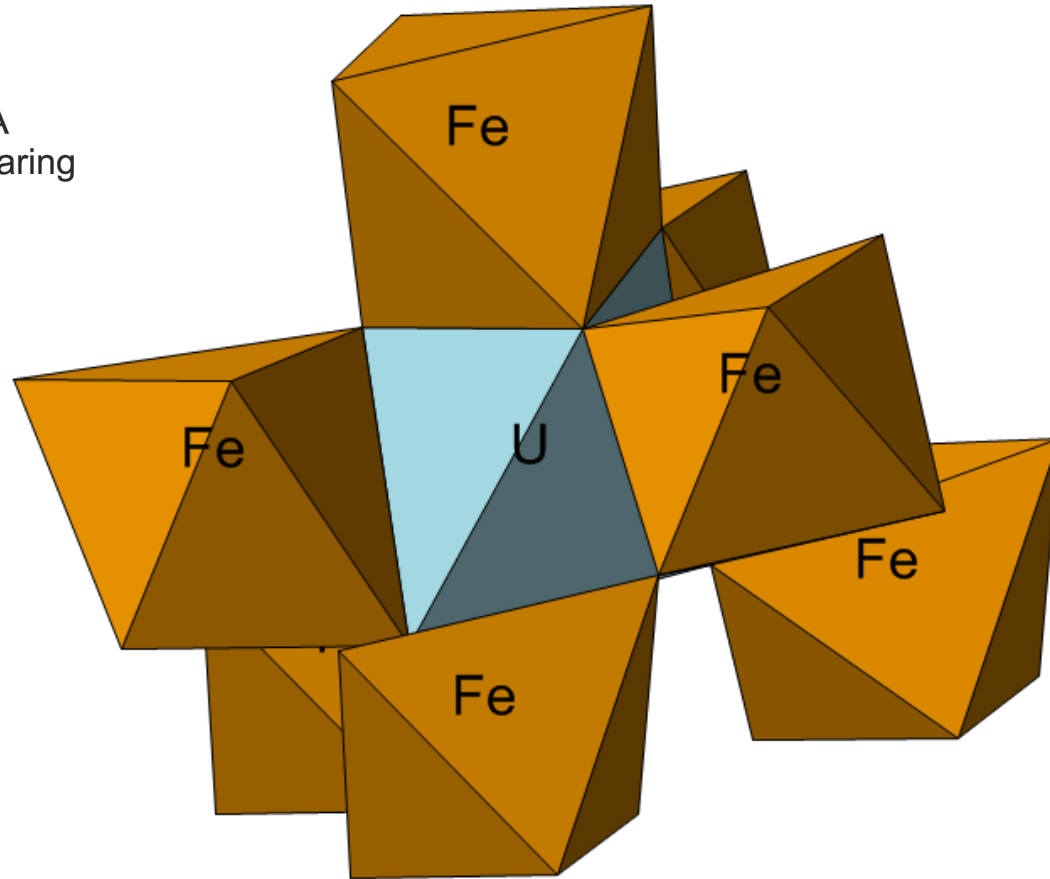
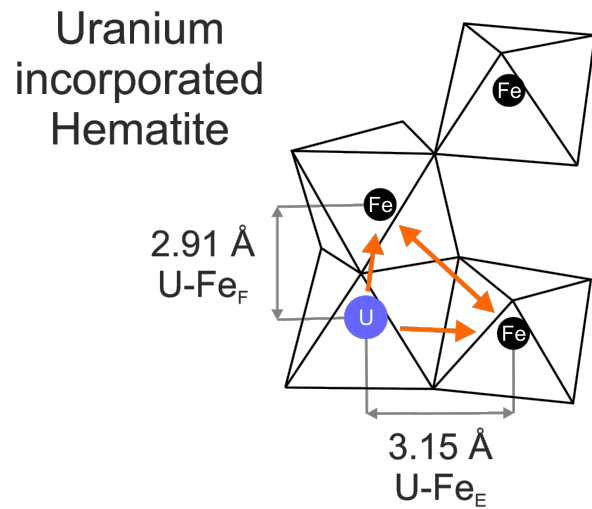
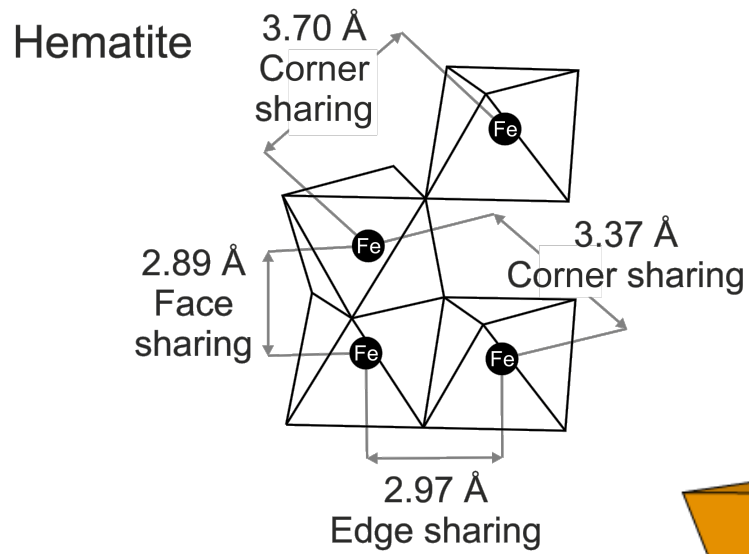
Magnesium silicates



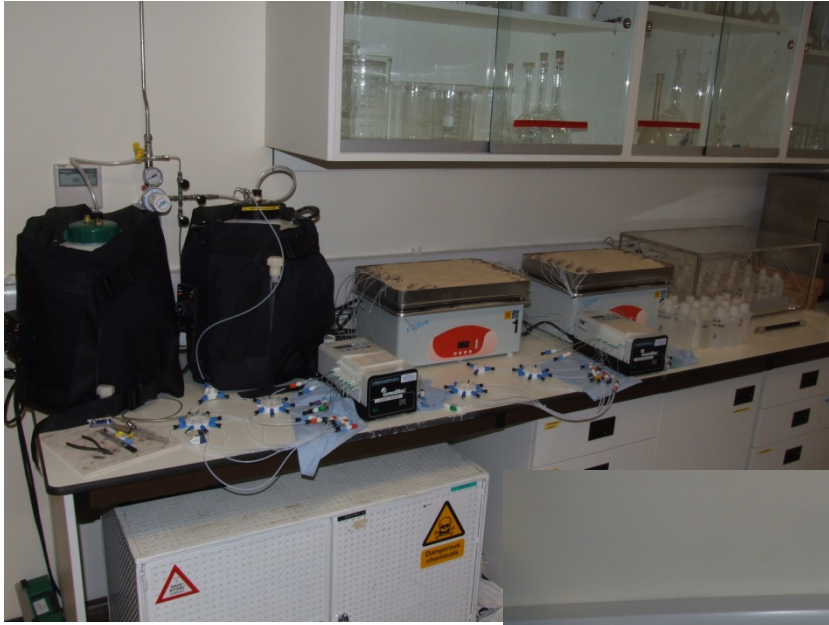
- Mg(Al)(K)-silicate 'sheet-like' phases on grain surfaces

- Mg-silicates suspended in solution

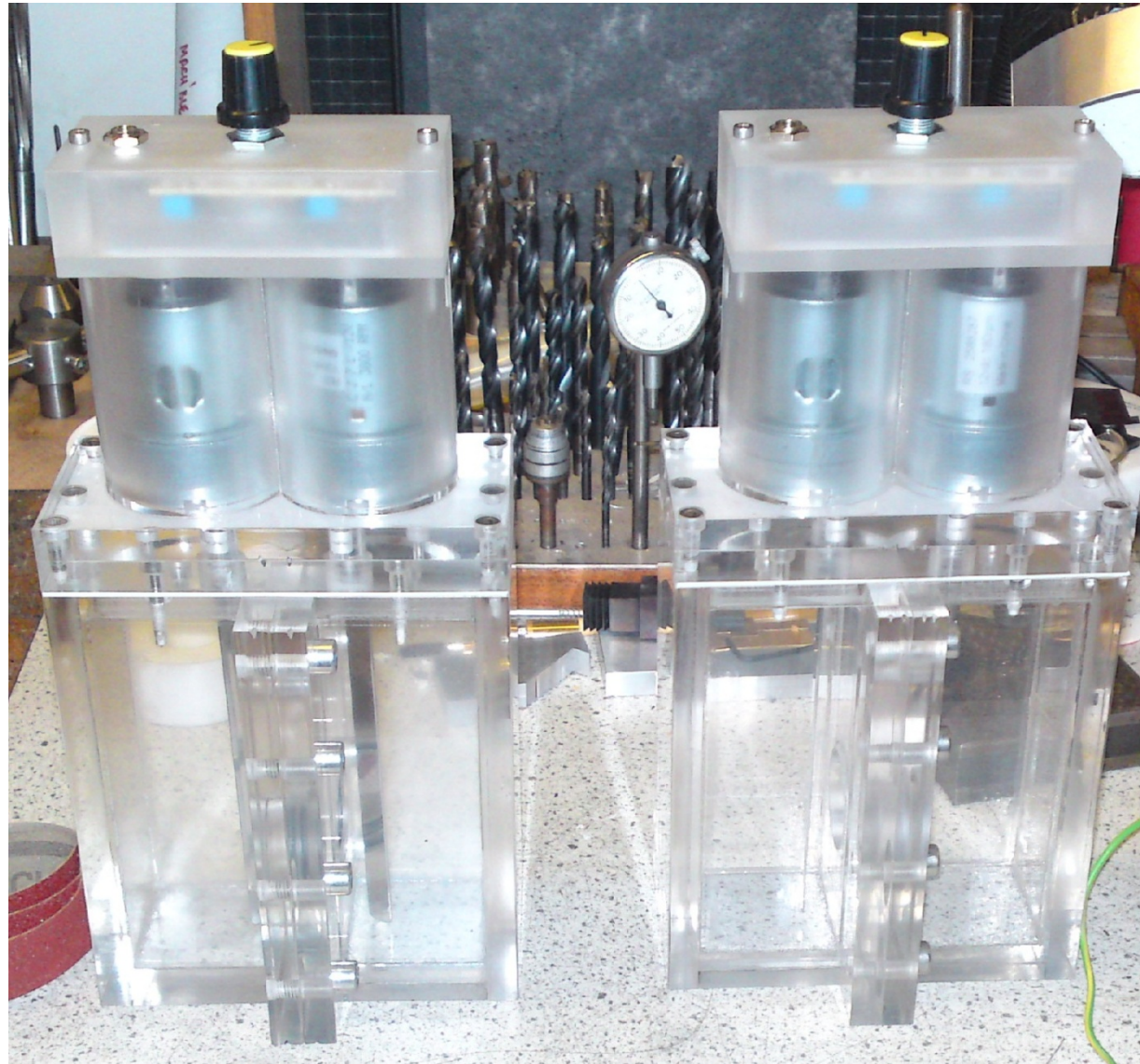
Hematite crystallisation at pH 10.5



adapted from Cornell and Schwertmann 2003



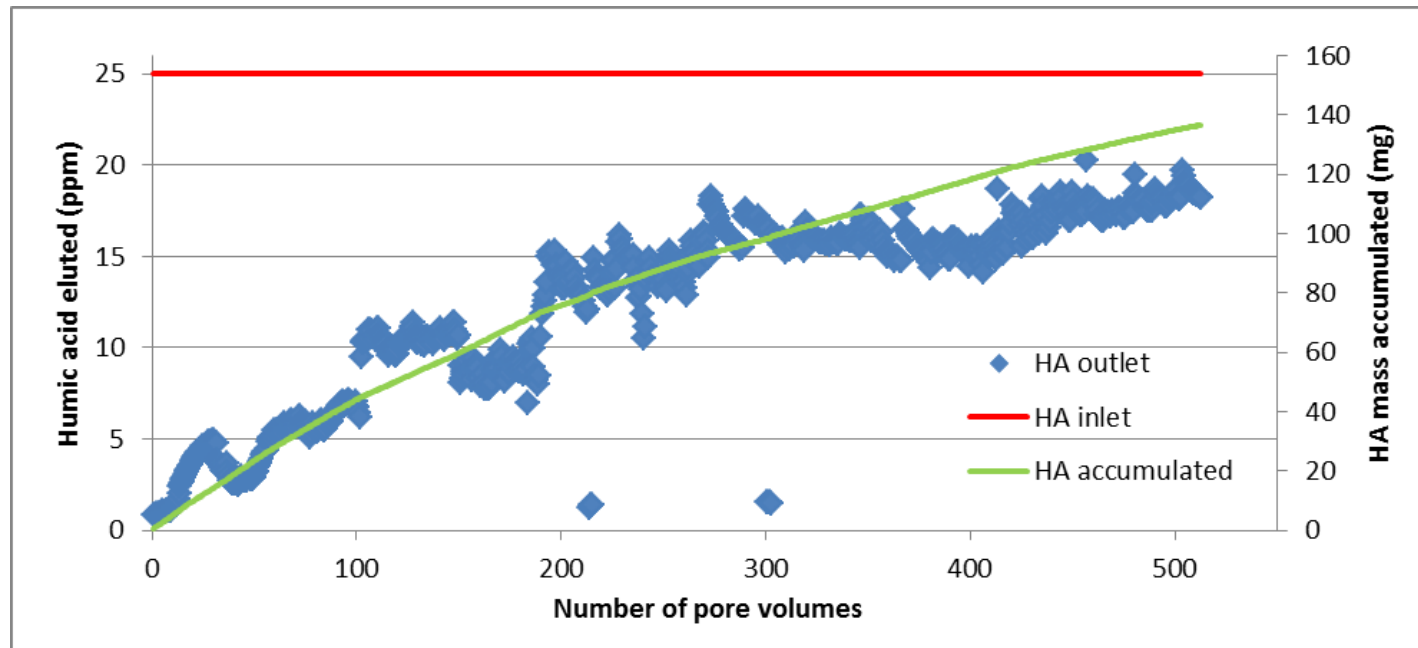
Kinetic experiments



Scenario III

Migration of U(VI) in YCL through sandstone saturated with HA

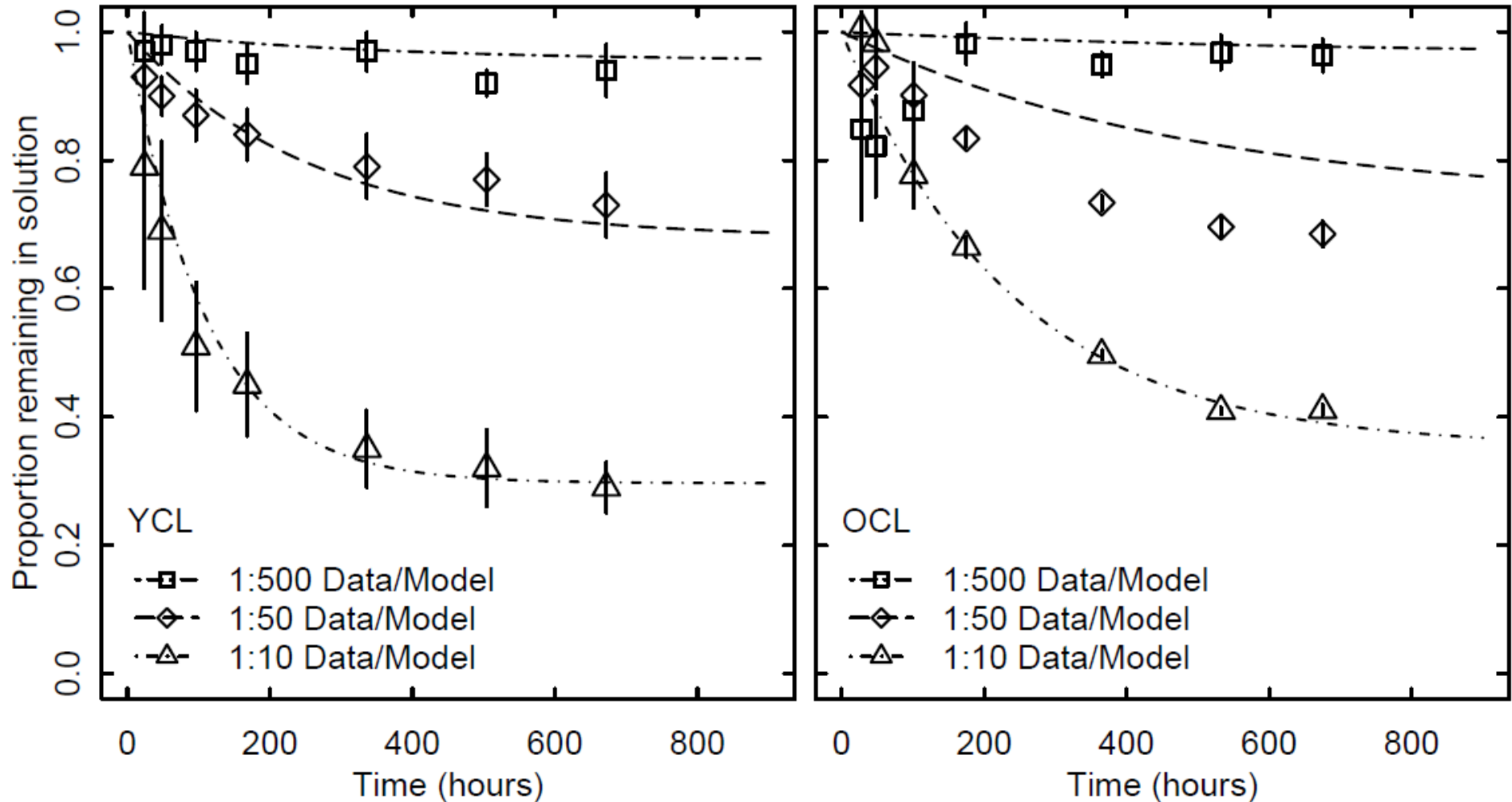
1st stage: saturation with HA under atmospheric conditions



◆ Flow 0.05 mL/min

◆ Concentration of HA = 25 ppm, 0.1 mol/L NaClO₄, pH 6.5

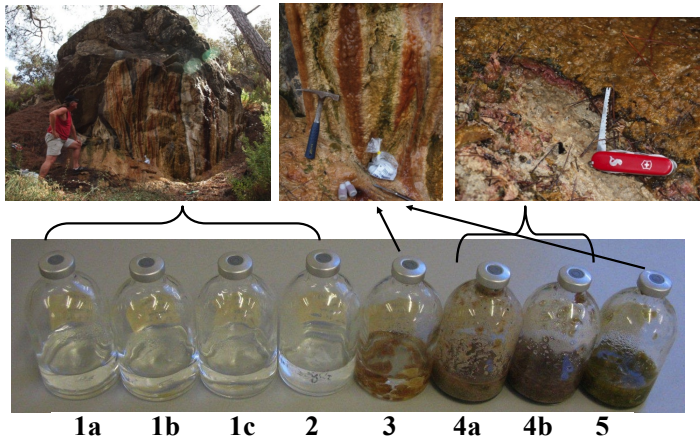
Low level sorption



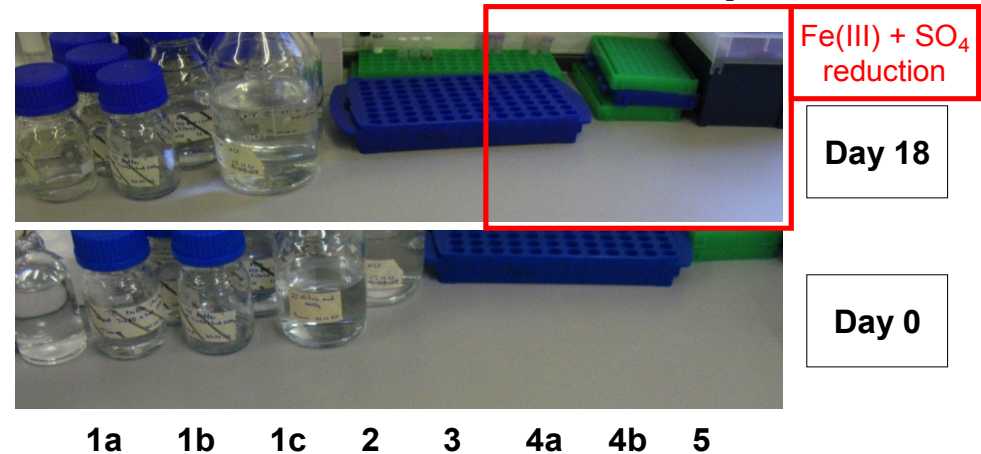
YCL rate constants of $k_1 = 1.78 \times 10^{-6} \text{ L mol}^{-1} \text{ s}^{-1}$ and $k_2 = 7.50 \times 10^{-7} \text{ s}^{-1}$.
OCL rate constants of $k_1 = 8.20 \times 10^{-7} \text{ L mol}^{-1} \text{ s}^{-1}$ and $k_2 = 5.12 \times 10^{-7} \text{ s}^{-1}$.
10 Bq ml⁻¹ ²³²U ($5.27 \times 10^{-11} \text{ M}$).

1 Microbial ecology of the Cyprus natural analogue

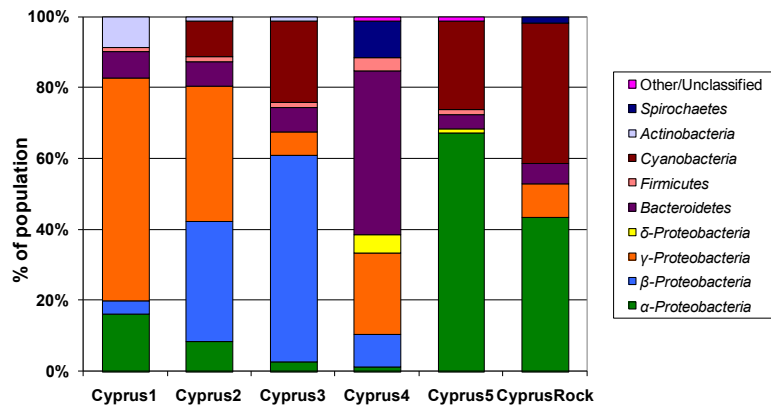
Experimental site & samples



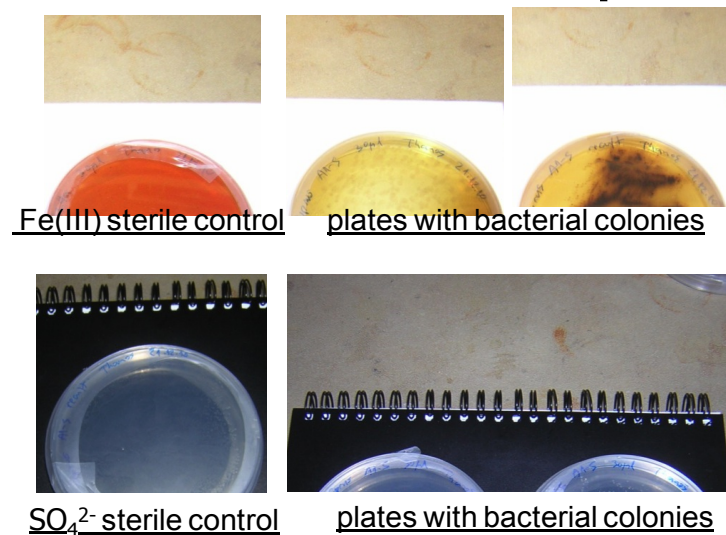
Enrichment cultures at pH 10



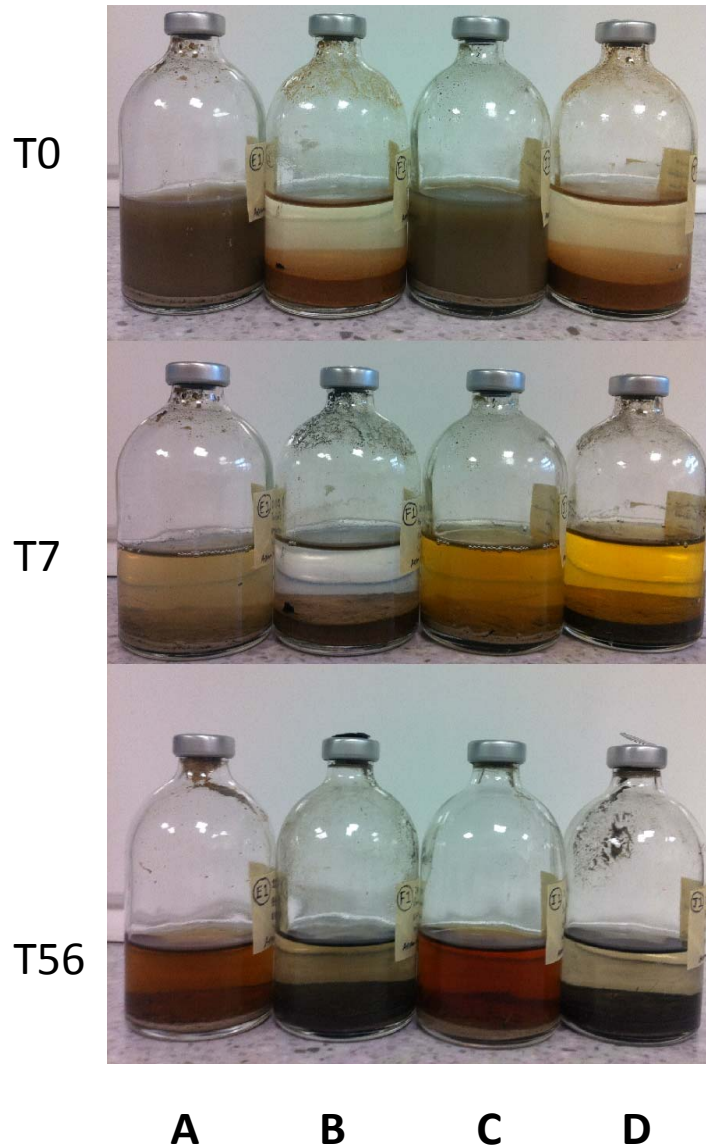
Bacterial diversity



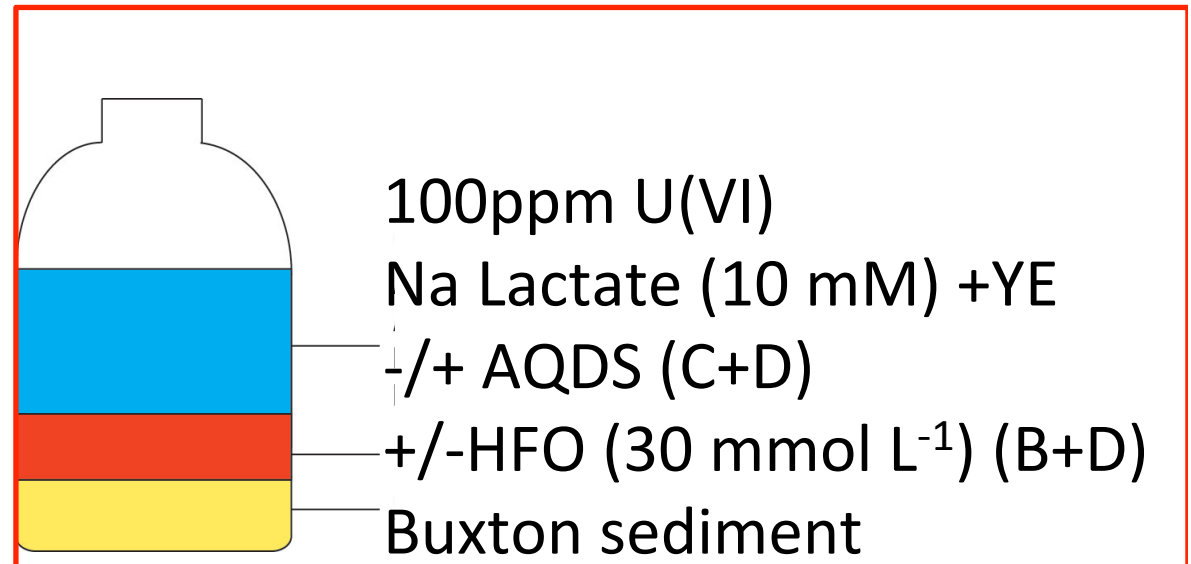
Isolated bacteria at pH 10



Active Batch Experiments

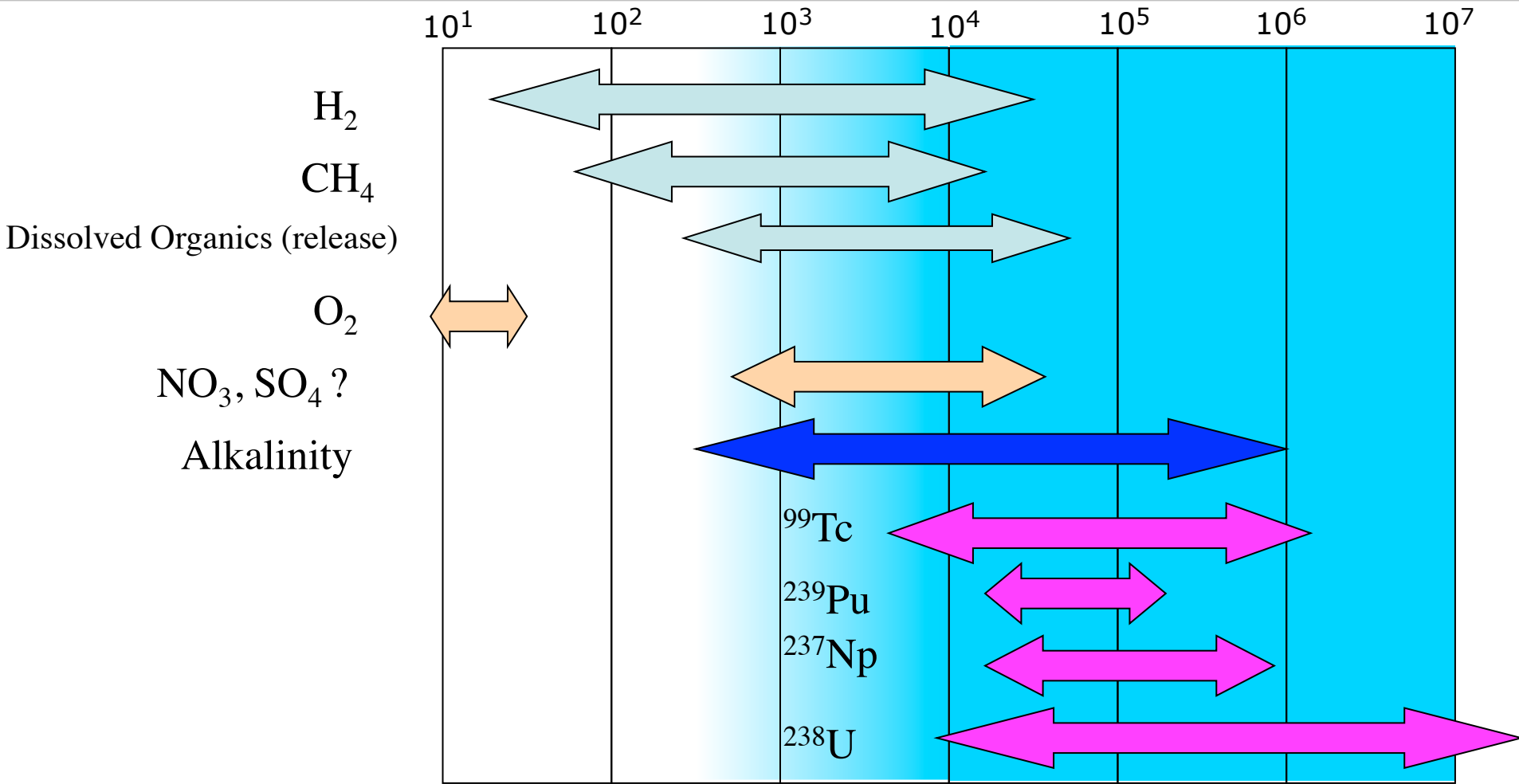


A= Bioreducing
 B=Bioreducing + HFO
 C= Bioreducing + AQDS
 D= Bioreducing + AQDS + HFO



XANES on all end members
EXAFS on post reduction mineral phases
Geochemical analyses (pH, Eh, Fe(II), U(VI))

Evolution of the CDZ (time (y) after waste emplacement)



Storage *wasteform/container degradation*

Gas Backfilling
ventilation

Re-saturation

Groundwater
advection, diffusion

Other Developments

Env-Rad-Net (Nov 2012)

- network in the multidisciplinary field of Environmental Radioactivity
- increasing application of STFC facilities to key research challenges
- exploiting synchrotron, neutron, laser and HPC approaches
- ***Sam Shaw and Gareth Law (Manchester)***

Royal Society Industrial Fellow (NNL to 2014)

- Microbial ecology of contaminated environments
- Radiation fluxes
- Bioremediation approaches
- ***Jon Lloyd (Manchester)***